

Table S3. Zero-point energies and electronic energies of reactants and transition state for the addition reaction of phenyl + acetylene with multi- and single-reference methods.

Multi-reference calculations				
Method		C ₆ H ₅ + C ₂ H ₂	Transition state	Barrier height (kcal/mol)
Zero-Point Energy (Hartree)				
CASSCF(11e,11o)/aug-cc-pVTZ		0.1201816	0.1203603	(13.15)
Single-Point Energy (Hartree)				
FIC-NEVPT2(11e,11o)				
cc-pVTZ		-308.1951293	-308.1905446	2.99
cc-pVQZ		-308.2891605	-308.2845441	3.01
CBS(cc-pVTZ, cc-pVQZ)		-308.3578033	-308.3531637	3.02
SC-NEVPT2(11e,11o)				
cc-pVTZ		-308.1936826	-308.1881823	3.56
cc-pVQZ		-308.2877437	-308.2820885	3.66
CBS(cc-pVTZ, cc-pVQZ)		-308.3312846	-308.3255577	3.73
MRCI(11e,11o)				
cc-pVDZ		-307.8170209	-307.8495567	-20.30
cc-pVTZ		-308.1016430	-308.1624778	-38.06
aug-cc-pVDZ		-307.8724421	-307.9366626	-40.19
CBS(cc-pVDZ, cc-pVTZ)		-308.2214688	-308.2942176	-45.54
Single-reference calculations				
Method	C ₆ H ₅	C ₂ H ₂	Transition state	Barrier height (kcal/mol)
Zero-Point Energy (Hartree)				
M06-2X/ 6-311+G(d,p)	0.0879820	0.0275260	0.1156490	(4.01)
Single-Point Energy (Hartree)				
B2PLYPD3/ def2-TZVPP	-231.4436114	-77.2938773	-308.7327118	3.08

Note:

(1) The active space (11e,11o) includes: (6e,6o) of C₆ ring, (2e,2o) of the nonreactive π -orbital of acetylene, (2e,2o) of the reactive π -orbital of acetylene, and (1e,1o) of the radical orbital.

(2) Electronic energies are refined using the FIC-NEVPT2 and SC-NEVPT2 methods with cc-pVTZ and cc-pVQZ basis sets and are then extrapolated to the complete basis set (CBS) limit by the formula Eq. (E1)

$$E_{\text{CBS}}^{\text{NEVPT2}} = E_{\text{cc-pVQZ}}^{\text{NEVPT2}} + (E_{\text{cc-pVQZ}}^{\text{NEVPT2}} - E_{\text{cc-pVTZ}}^{\text{NEVPT2}}) \times 3^3 / (4^3 - 3^3) \quad (\text{E1})$$

(3) Electronic energies are refined using the MRCI method with cc-pVDZ and cc-pVTZ basis sets and are then extrapolated to the complete basis set (CBS) limit by the formula Eq. (E2)

$$E_{\text{CBS}}^{\text{MRCI}} = E_{\text{cc-pVTZ}}^{\text{MRCI}} + (E_{\text{cc-pVTZ}}^{\text{MRCI}} - E_{\text{cc-pVDZ}}^{\text{MRCI}}) \times 2^3 / (3^3 - 2^3) \quad (\text{E2})$$