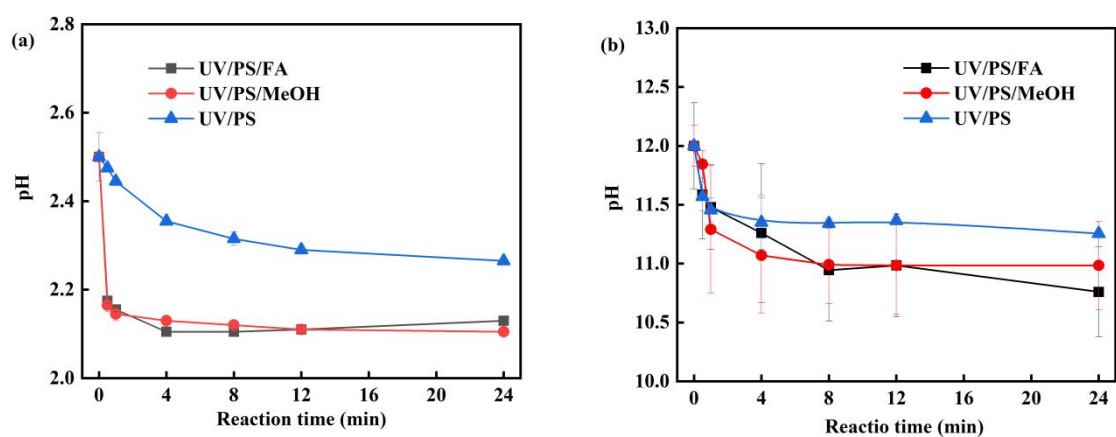
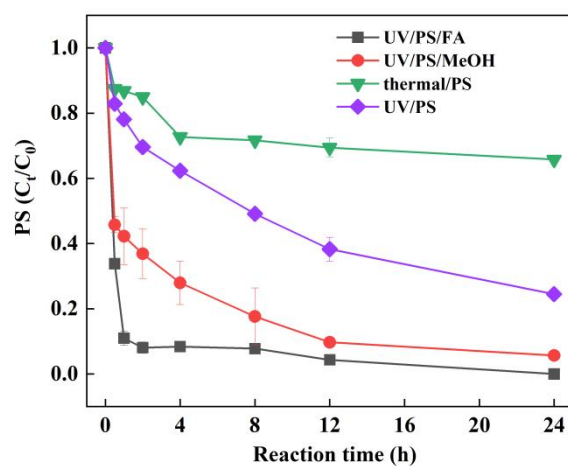


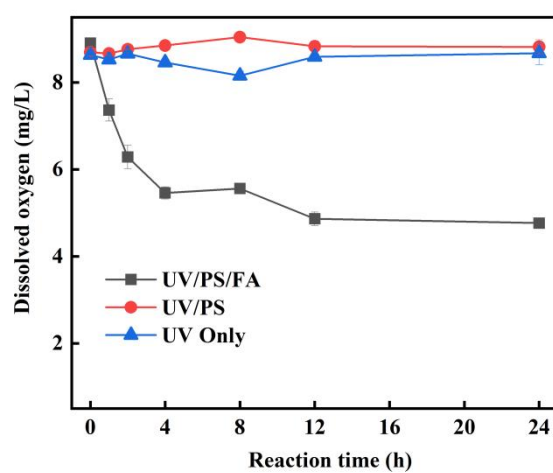
**Figure S1** The effect of FA concentration on the defluorination of PFOA by the UV/H<sub>2</sub>O<sub>2</sub>/FA system. Reaction conditions: [PFOA] = 20  $\mu$ M, [H<sub>2</sub>O<sub>2</sub>] = 4 mM. [FA] = 2-20 mM, initial pH of 2.5, anaerobic environment.



**Figure S2** Variations of solution pH with reaction time under different initial pH conditions. Reaction conditions: [PFOA] = 20  $\mu$ M, [PS] = 4 mM, [FA] = 2 mM, (a) initial pH of 2.5, (b) initial pH of 12, anaerobic environment.



**Figure S3** Decomposition of PS in PS activation systems in the presence or absence of FA/MeOH. Reaction conditions: [PFOA] = 20  $\mu$ M, [PS] = 4 mM, [FA] = 2 mM, [MeOH] = 2 mM, initial pH of 2.5, anaerobic environment, 80  $^{\circ}$ C for the thermal/PS system, 20  $^{\circ}$ C for the rest of the systems.



**Figure S4** Variations of dissolved oxygen in UV/PS/FA, UV/PS, and UV systems.

Reaction conditions: [PFOA] = 20  $\mu$ M, [PS] = 4 mM, [FA] = 2 mM, initial pH of 2.5, initial DO of 8.5 mg/L.